

# ***Chemistry II (Organic)***

## ***Heteroaromatic Chemistry***

### ***LECTURE 1***

#### ***Introduction & overview***

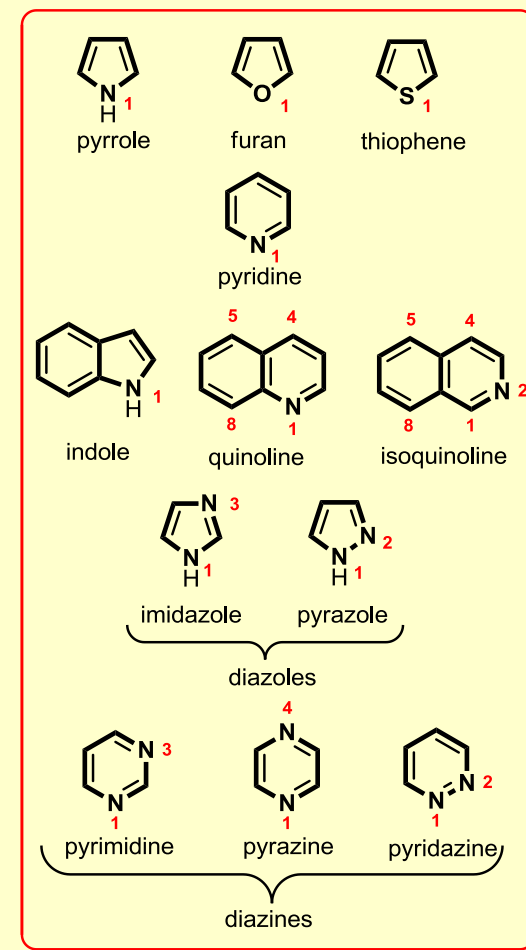
Alan C. Spivey  
a.c.spivey@imperial.ac.uk

**Imperial College**  
**London**

***Feb 2012***

# Format & scope of lecture 1

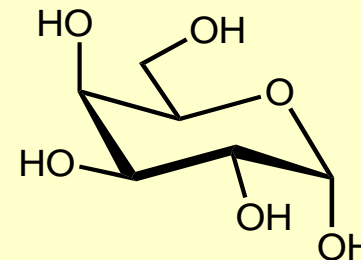
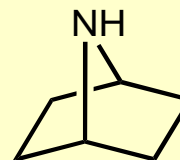
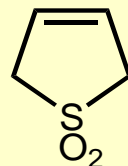
- **Definitions of structural types**
  - heterocyclic compounds
  - aromatic heterocycles (heteroaromatics)
- **Occurrence and relevance**
  - serotonin & histamine
  - LSD
  - cimetidine (Tagamet™) & ranitidine (Zantac™)
  - atorvastatin calcium (Lipitor™) & torcetrapib
  - sildenafil citrate (Viagra™) & vardenafil citrate (Levitra™)
  - celecoxib (Celebrex™) & rofecoxib (Vioxx™)
  - natural products & chiral auxiliaries/catalysts
  - agrochemicals & ‘smart’ materials
- **Physical, chemical & spectroscopic properties**
  - recap of structure of benzene and aromaticity
  - valence bond and molecular orbital representations
  - Hückel’s rule
  - resonance energies and symmetry



# Definitions of Structural Types

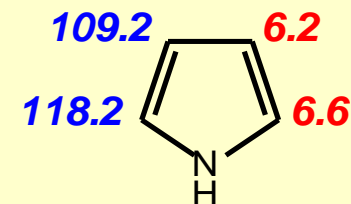
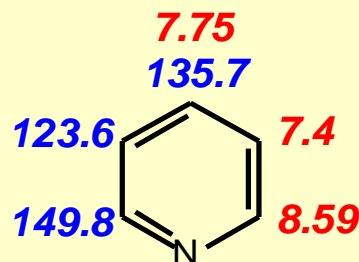
## ■ Heterocyclic Compounds

- composed of primarily carbon atoms
- contain at least one other element
- **nitrogen, oxygen** and **sulphur**
- **half** the known organic compounds
- present in almost ALL **pharmaceuticals** and **agrochemicals**



## ■ AROMATIC heterocycles (HETEROAROMATICS)

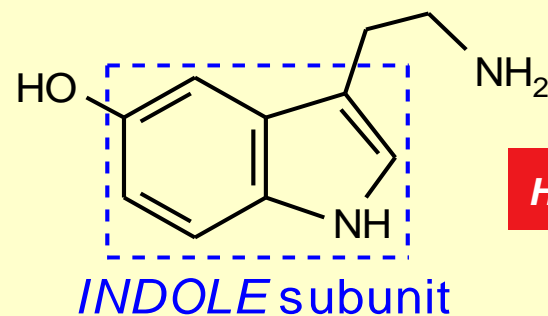
- obey **HÜCKEL'S RULE** (see later)
  - **~planar**
  - **contiguous, cyclic array of p-orbitals (delocalised electrons)**
  - **4n+2** electrons delocalised (n is an integer)
- possess characteristics of arenes
  - 'anomalous' chemical properties
  - **unusual NMR chemical shifts**
  - occur widely in nature



<sup>13</sup>C and <sup>1</sup>H chemical shifts (δ/ppm)

## ■ 5-HYDROXYTRYPTAMINE (SEROTONIN)

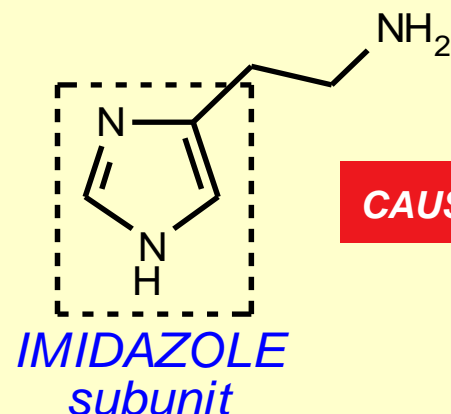
- ❑ first isolated in 1948
- ❑ possesses wide range of pharmacological activity:
  - ❑ induces arterial constriction
  - ❑ affects mood and appetite
  - ❑ induces blood platelet aggregation



**HALLUCINOGEN**

## ■ HISTAMINE

- ❑ isolated in 1927
- ❑ derived in vivo from decarboxylation of amino acid histidine
  - ❑ liberated from cells upon injury
  - ❑ lowers blood pressure
  - ❑ directly involved in allergic reactions
- ❑ **ANTIHISTAMINES** widely used therapeutically (see later)

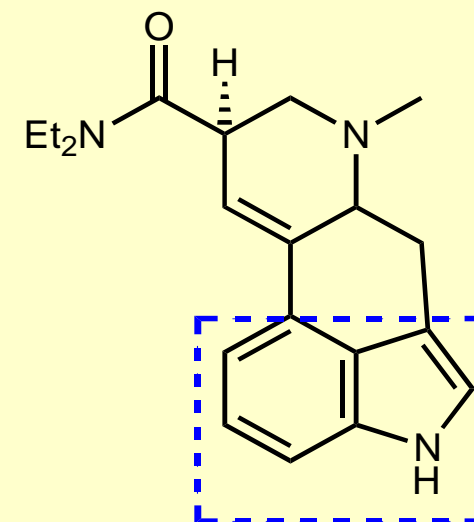


**CAUSES INFLAMMATION**

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## ■ **LYSERGIC ACID DIETHYLAMIDE (LSD)**

- ❑ derivative of Lysergic Acid
  - found in **ergot**, a common grain fungus
- ❑ many potent physiological effects documented
- ❑ LSD 1<sup>st</sup> made in 1938 by Dr. Albert Hoffmann (Sandoz, Basel, Switzerland)
  - spring 1943, Hofmann resynthesized 'LSD-25'
  - cycled home after self-dosing: suffered 'trip'
- ❑ known as 'acid'
- ❑ strongly hallucinogenic/psychedelic
  - induces enhanced sensory perception
- ❑ many adverse reactions
  - 'bad trip'
  - flashbacks may occur years later
  - can induce permanent psychosis
  - has serotonin-blocking effect
  - interferes with dopamine action

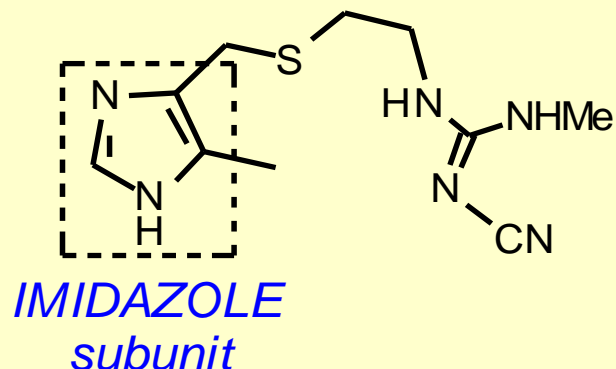


*INDOLE subunit*



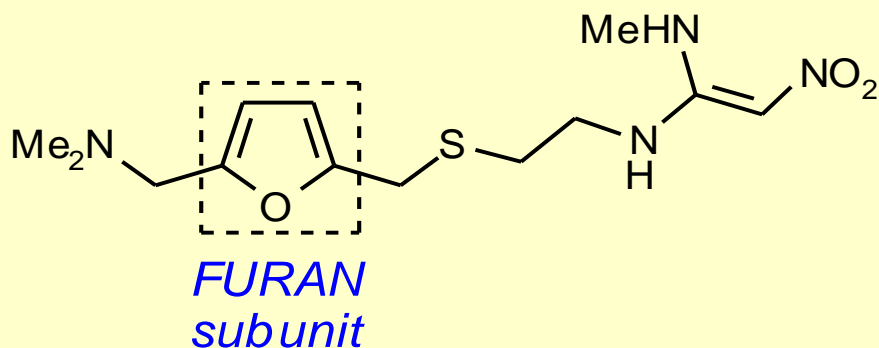
Prof. Albert Hofmann 100<sup>th</sup> birthday in 2006  
I. Amato 'Trip of a century' *Chem. Eng. News.* **2006**, Feb, 43 ([DOI](#))  
D. Nichols 'LSD: cultural revolution & medical advances'  
*Chem. World* **2006**, Jan, 30 ([DOI](#))

## ■ CIMETIDINE (Tagamet®)



- SmithKlineFrench (now GSK) **histamine  $H_2$ -receptor antagonist**
- blocks acid secretions
- Indicated for:
  - Treatment & prevention of stomach ulcers
  - gastrooesophageal reflux disorder (GERD)
- World Health Organization: “one of world’s most essential drugs”

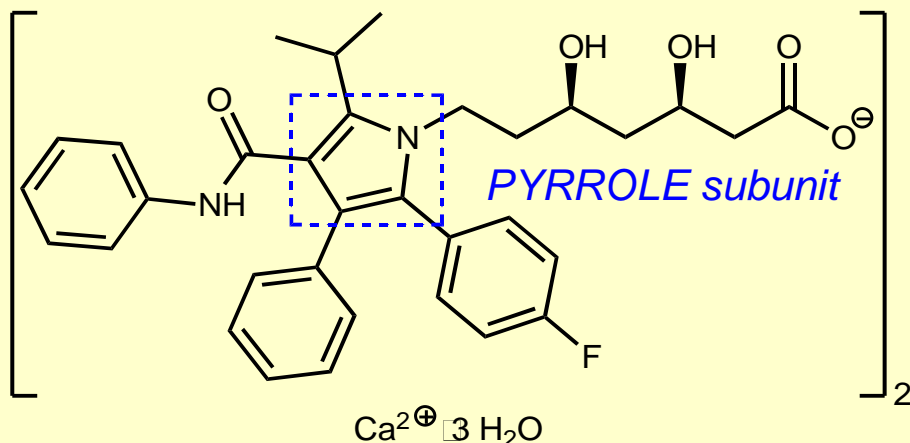
## ■ RANITIDINE (Zantac®)



- GlaxoWellcome (now GSK) **histamine  $H_2$ -receptor antagonist**
- blocks acid secretions
- Indicated for:
  - Treatment & prevention of stomach ulcers
  - gastrooesophageal reflux disorder (GERD)
- World Health Organization: “one of world’s most essential drugs”

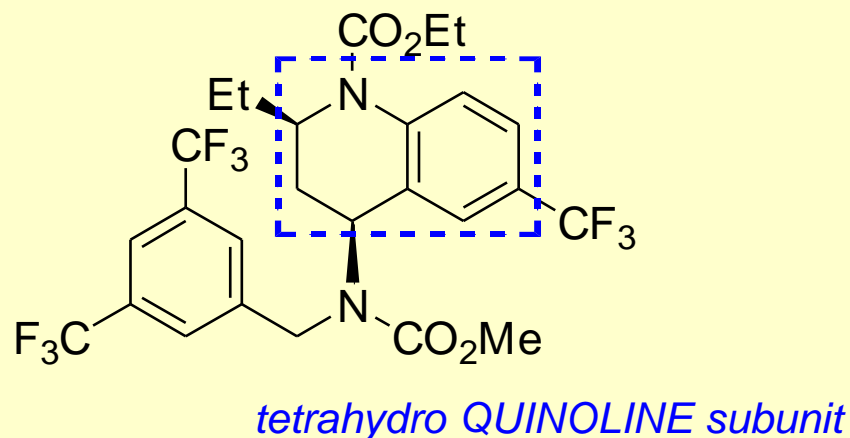
# Heteroaromatic Pharmaceuticals – Cholesterol-lowering Drugs (Statins & CETP inhibitors)

## ■ ATORVASTATIN CALCIUM (Lipitor®)



- Pfizer **cholesterol-lowering statin**
- HMGCoA inhibitor (see next slide)
- indicated for stroke and heart-attack prevention
- first drug to reach **>\$10 billion** annual sales (2004: \$10.8 bn)
- “the most popular drug ever sold ever in the history of the world”

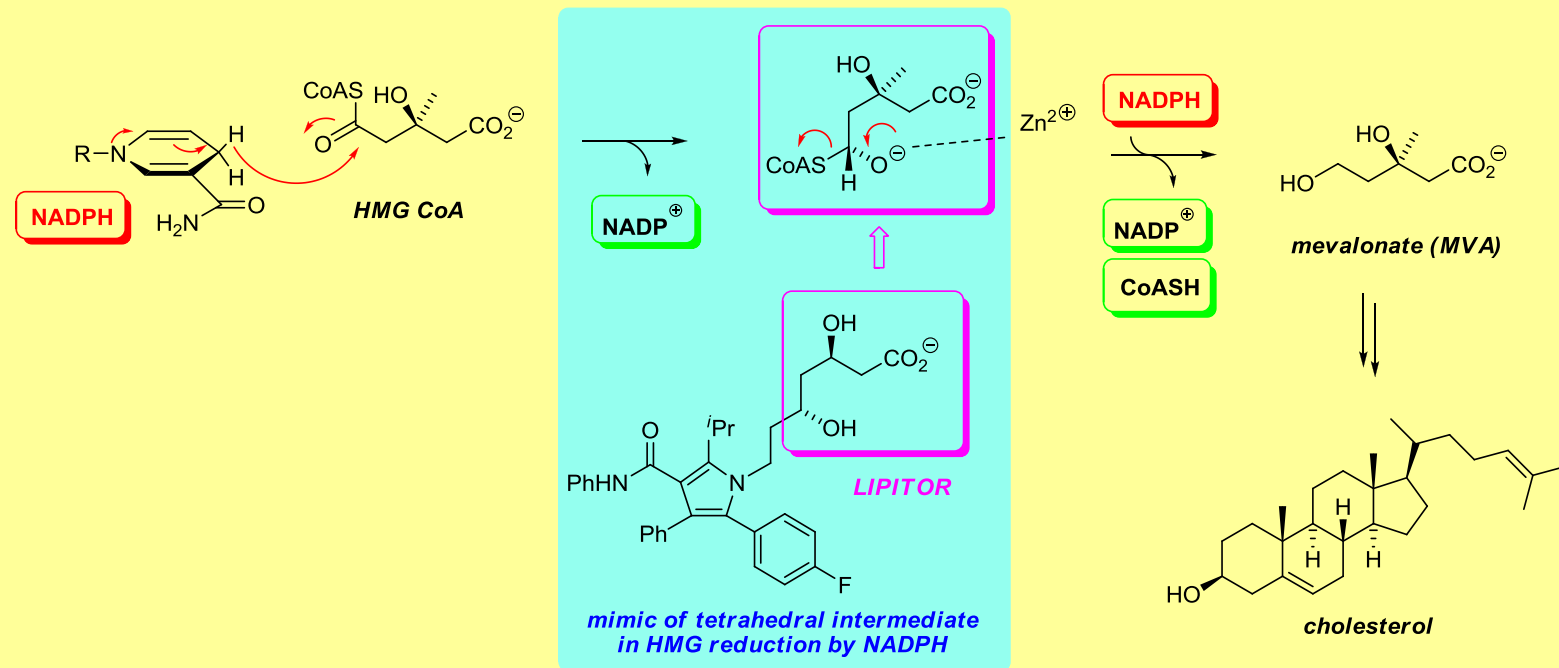
## ■ TORCETRAPIB (CP-529414)



- Pfizer **cholesteryl ester transfer protein (CETP) inhibitor**
- designed to reduce build-up of low-density lipoproteins (LDLs) ‘bad cholesterol’ as a co-therapy with Lipitor for cardiovascular protection
- **Nov 30<sup>th</sup> 2006:** Pfizer’s chief executive “This will be one of the most important compounds of our generation.”
- **Dec 2<sup>nd</sup> 2006:** phase III trials abandoned ([DOI](#))

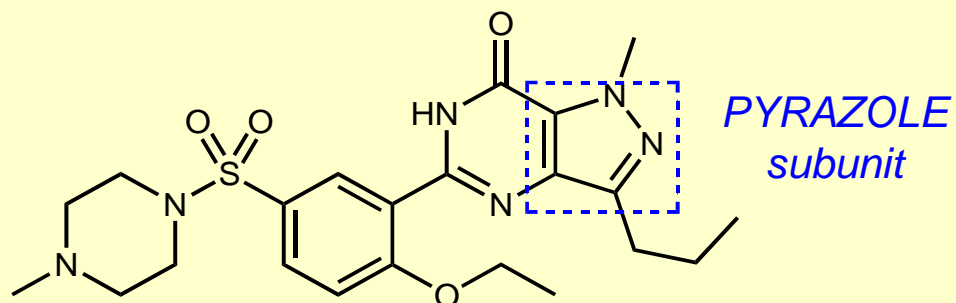
# HMG CoA reductase inhibitors - Statins

- **HMG CoA** → **MVA** is the **rate determining step** in the biosynthetic pathway to **cholesterol**
  - 33 enzyme mediated steps are required to biosynthesise cholesterol from acetyl CoA & in principle the inhibition of any one of these will serve to break the chain. In practice, control rests with HMG-CoA reductase as the result of a variety of biochemical feedback mechanisms
- ‘**Statins**’ inhibit HMG CoA reductase and are used clinically to treat **hypercholesterolemia** - a causative factor in **heart disease**
  - e.g. **Lipitor™** (Pfizer) - a competitive inhibitor of HMG-CoA reductase



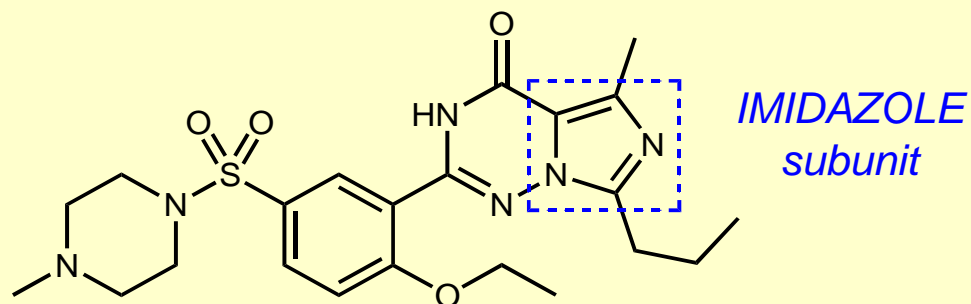


## ■ **SILDENAFIL CITRATE (Viagra®)**



- ❑ Pfizer **type 5 phosphodiesterase (PDE) inhibitor**
- ❑ prevents breakdown of cyclic GMP & originally designed for treatment of hypertension (high blood pressure)
- ❑ indicated for treatment for erectile dysfunction
- ❑ annual sales: **\$1.7 bn**

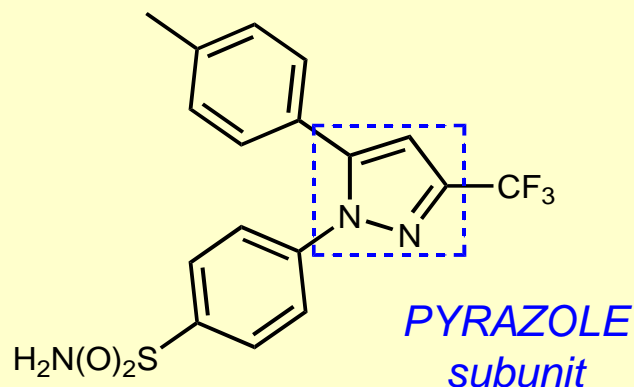
## ■ **VARDENAFIL CITRATE (Levitra®)**



- ❑ Bayer **type 5 phosphodiesterase (PDE) inhibitor**
- ❑ prevents breakdown of cyclic GMP
- ❑ Indicated for treatment for erectile dysfunction
- ❑ annual sales: **\$0.25 bn**

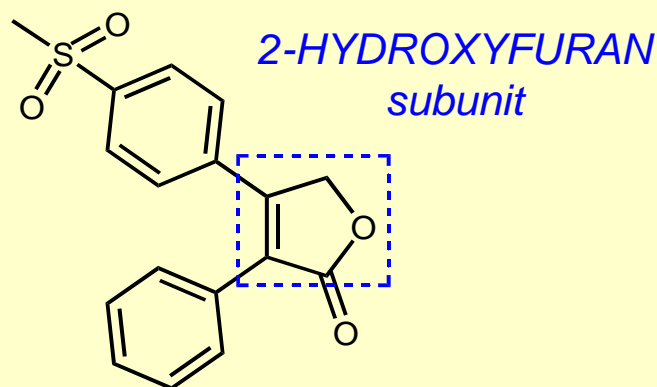
# Heterocyclic Pharmaceuticals – Anti-inflammatory Drugs

## ■ CELECOXIB (Celebrex®)



- Pfizer *non-steroidal anti-inflammatory drug (NSAID)*
- selective COX-2 inhibitor (see next slide)
- indicated for:
  - osteoarthritis
  - rheumatoid arthritis
  - acute pain
- US sales 2004: **\$2.1 bn**

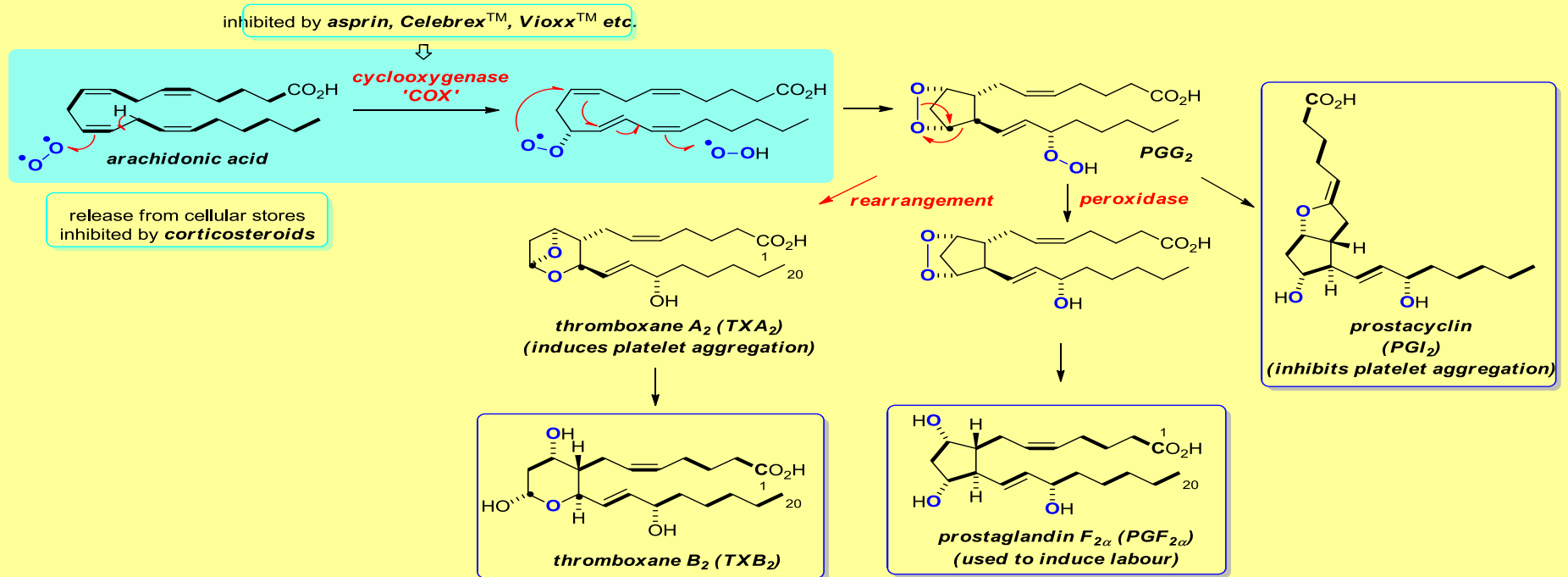
## ■ ROFECOXIB (Vioxx®)



- Merck *non-steroidal anti-inflammatory drug (NSAID)*
- selective COX-2 inhibitor (see next slide)
- indicated for:
  - osteoarthritis
  - rheumatoid arthritis
  - acute pain
- US sales 2004: **\$0.9 bn**
- withdrawn from market December 2004 due to cases of heart attack

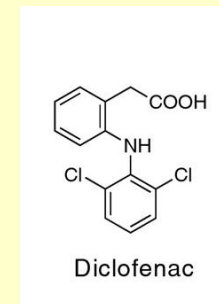
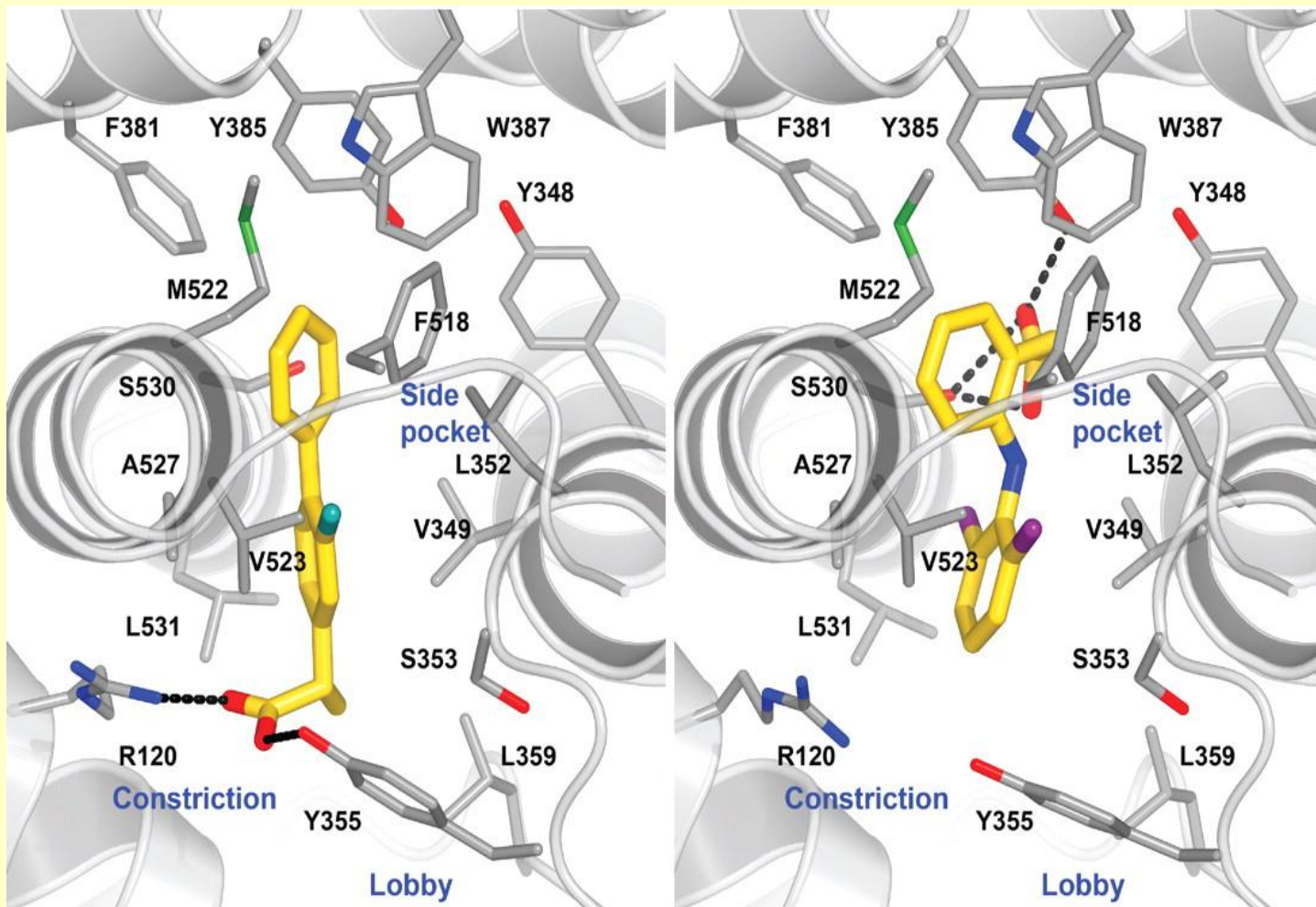
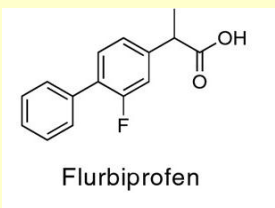
# Biosynthesis of Prostaglandins & Thromboxanes

- prostaglandins & thromboxanes **are derived from further oxidative processing of arachidonic acid**
- **both are important** hormones **which control e.g. smooth muscle contractility (blood pressure), gastric secretion, platelet aggregation & inflammation (< nM activity)**
  - various pharmaceuticals including **corticosteroids, NSAIDs & aspirin** inhibit biosynthetic steps in these pathways



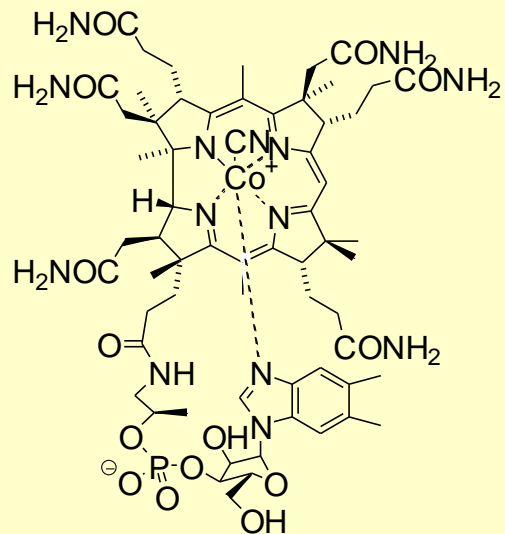
# Crystal structures of flurbiprofen (Ansaid®) & diclofenac (Zolterol®) bound in mCOX-2 active site

■ Duggan *et al. J. Biol. Chem.* **2010**, 285, 34950-34959 ([DOI](#))

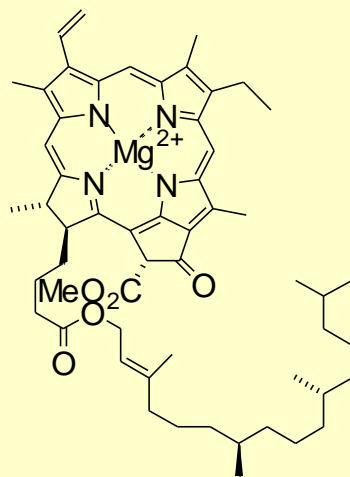


# Other Examples of Important Heterocyclic Compounds

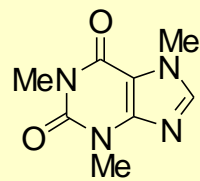
## More bioactive natural products



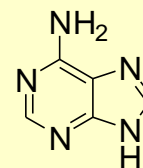
vitamin B<sub>12</sub>



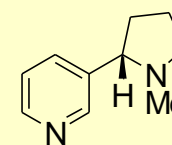
chlorophyll



caffeine

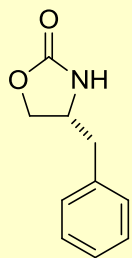


adenine (ATP, DNA..)

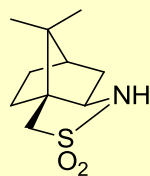


nicotine

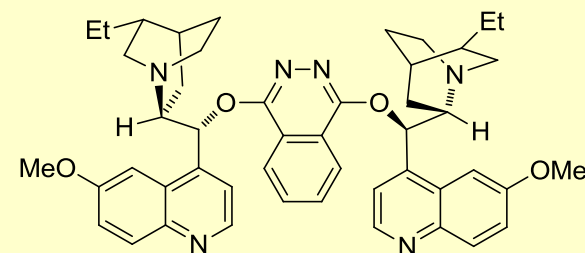
## Chiral auxiliaries and catalysts



Evans' oxazolidinone



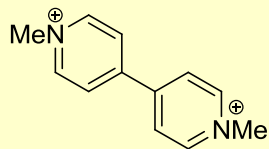
Oppolzer's camphor sultam



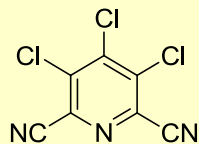
(DHQ)<sub>2</sub>PHAL (Sharpless' AD catalyst)

# Other Examples of Important Heterocyclic Compounds

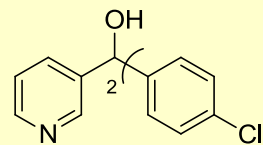
## ■ Agrochemicals



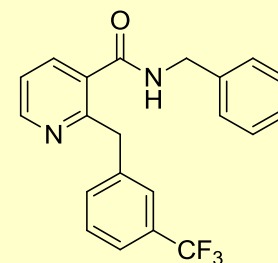
**paraquat**  
(weed-killer)



**dowco 263**  
(fungicide)

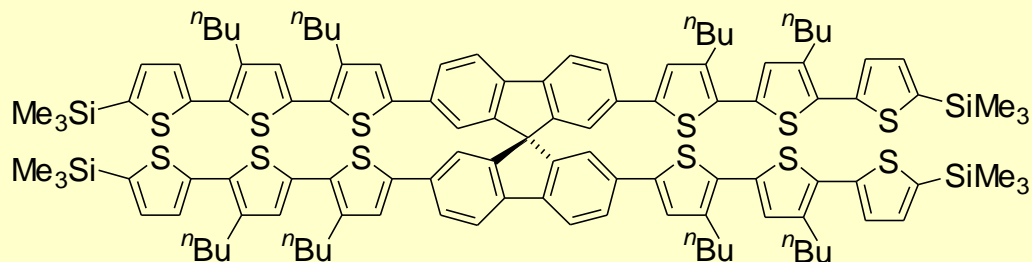


**parinol**  
(fungicide)

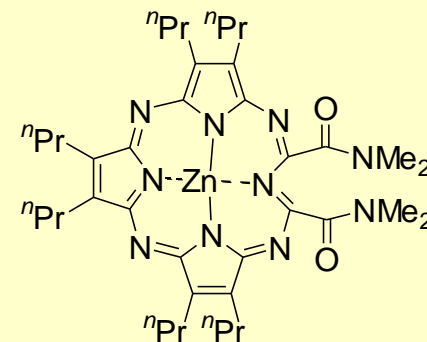


**phenoxynicotinamide**  
(herbicide)

## ■ 'Smart' materials



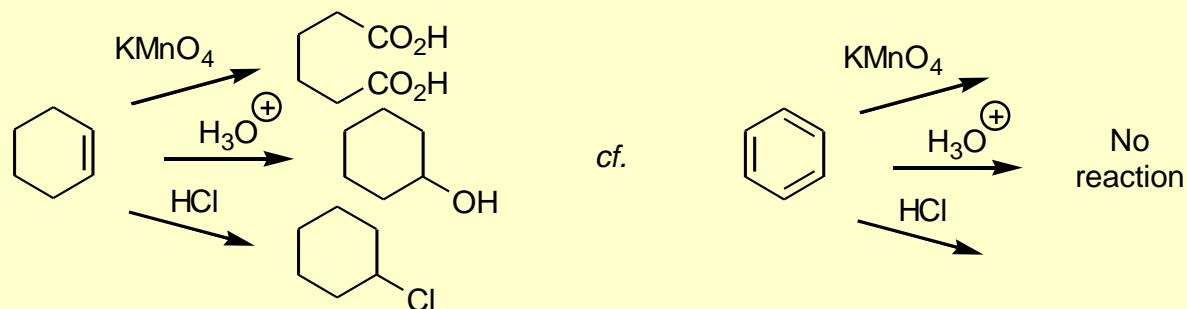
An orthogonally fused conjugated oligomer comprised of thiophene units  
(a potential molecular-scale electronic device)



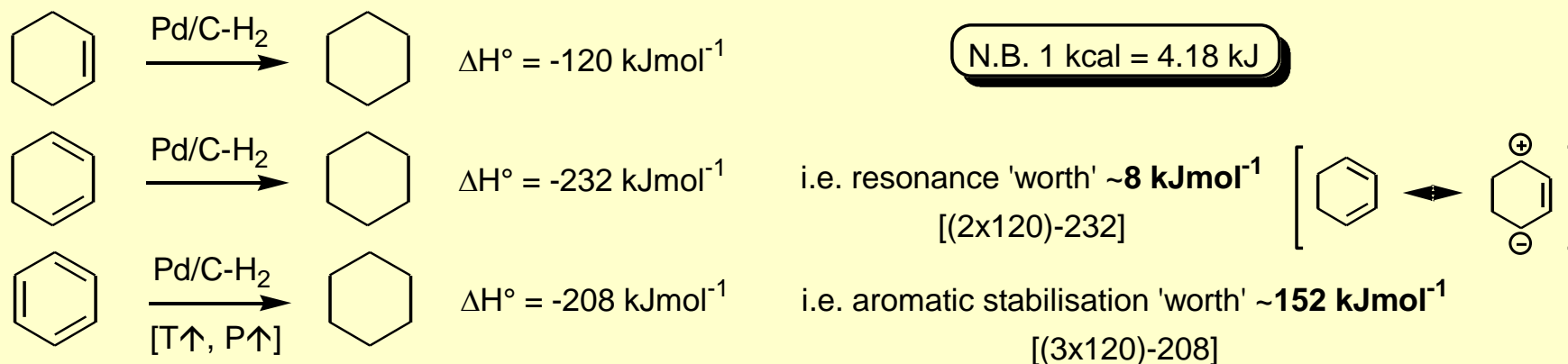
**seco-porphyrazine**  
(<sup>1</sup>O<sub>2</sub> photosensitiser for photodynamic chemotherapy)

# Benzene – Stability and Resonance Energy

- Benzene is unusually unreactive (*i.e.* stable) *cf.* alkenes



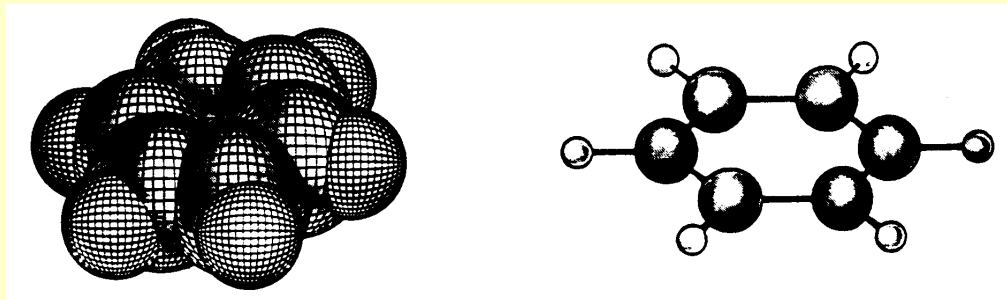
- A direct estimate of resonance energy can be made by consideration of heats of hydrogenation of cyclohexenes



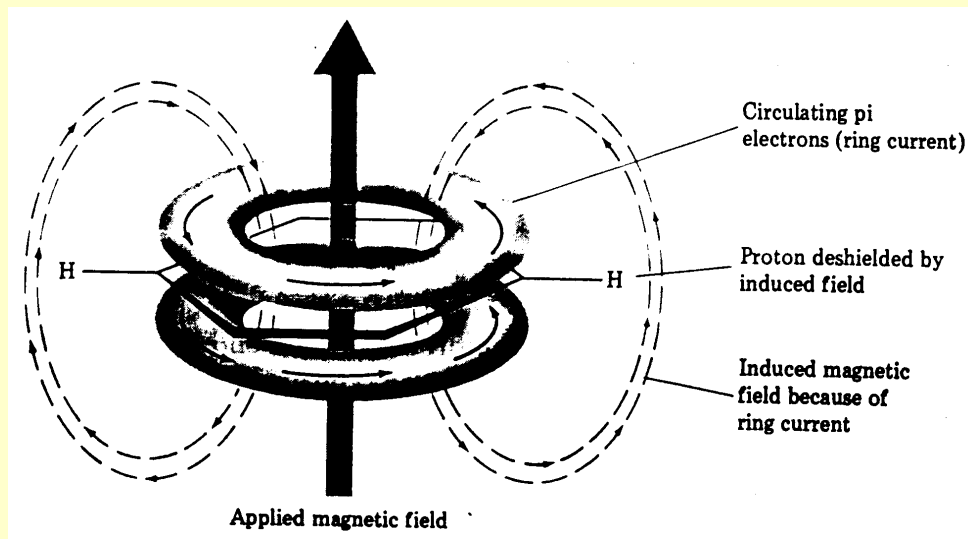
□ see: K.P.C.Vollhardt *J. Am. Chem. Soc.* **2000**, 122, 7819 ([DOI](#))

## Benzene – Structure and NMR Properties

- X-ray data show all bond lengths to be the same (1.39 Å *cf.* average C-C 1.54 Å & average C=C 1.34 Å)



- $^1\text{H}$  NMR reveals aryl protons experience deshielding (*i.e.* → low field) – due to induced ring current:



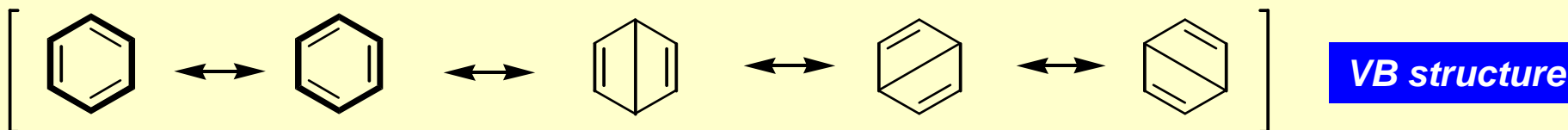


# Benzene – Valence Bond and Molecular Orbital Representations of Aromaticity

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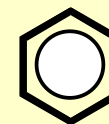
## ■ VALENCE BOND (VB) THEORY

- ❑ resonance hybrids - imaginary structures which differ only in position of electrons (atoms/nuclei do not move)
- ❑ Not all resonance structures contribute equally - 'real' structure is weighted average of resonance structures



## ■ MOLECULAR ORBITAL (MO) THEORY

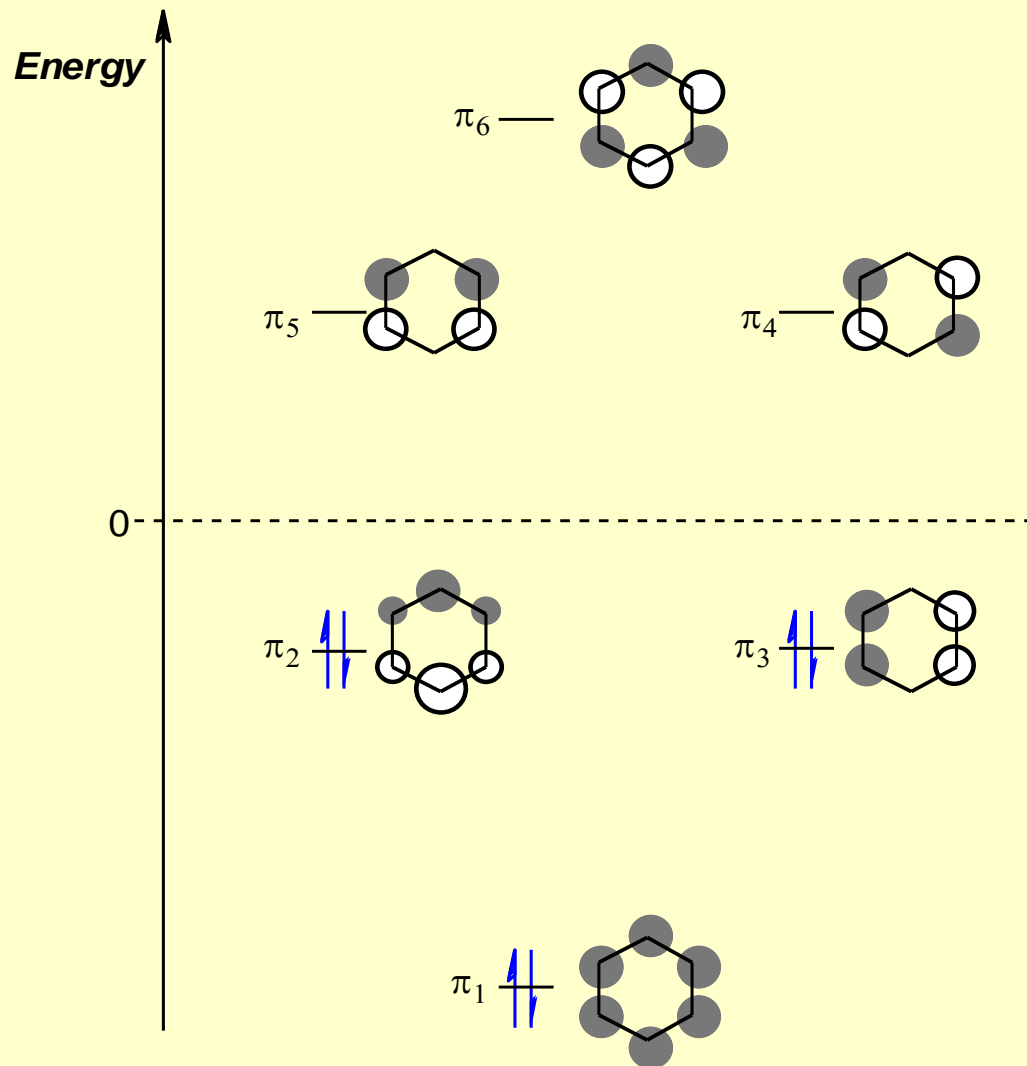
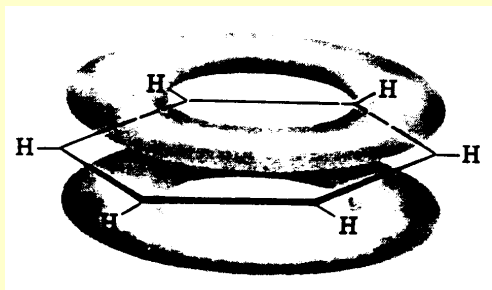
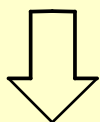
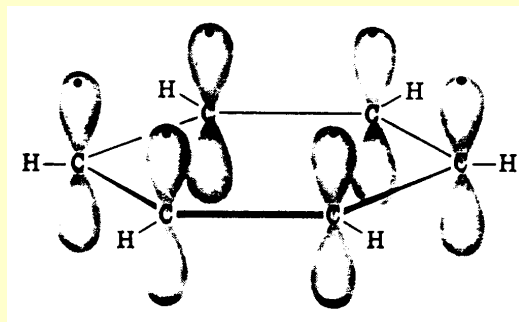
- ❑ Linear Combination of Atomic Orbitals (LCAO)
- ❑  $\sigma$ -bonding framework formed from  $sp^2$  hybridised carbons
- ❑ leaves p-orbital on each C atom *orthogonal* to ring
- ❑ 6 atomic p-orbitals (AOs)  $\rightarrow$  [LCAO maths]  $\rightarrow$  6 MOs
- ❑ each MO capable of containing 2 electrons
- ❑ 6 electrons available to occupy the 6 MOs
- ❑ placed in 3 molecular orbitals of lowest energy: **bonding orbitals**
- ❑ 3 **anti-bonding orbitals** remain vacant



**MO structure**

# Benzene - Molecular Orbital Description

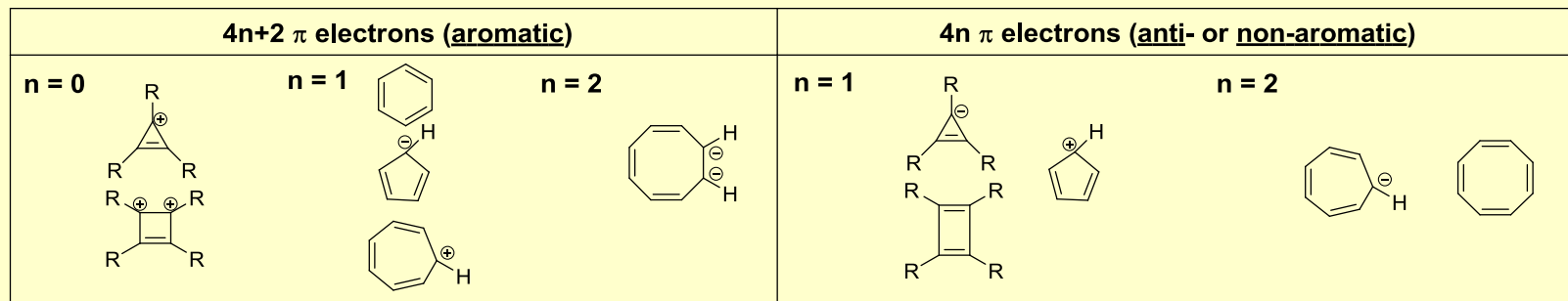
- Molecular orbital theory rationalises reactions and properties of benzene:



# Benzene – Hückel's Rule and Musulin-Frost Diagrams

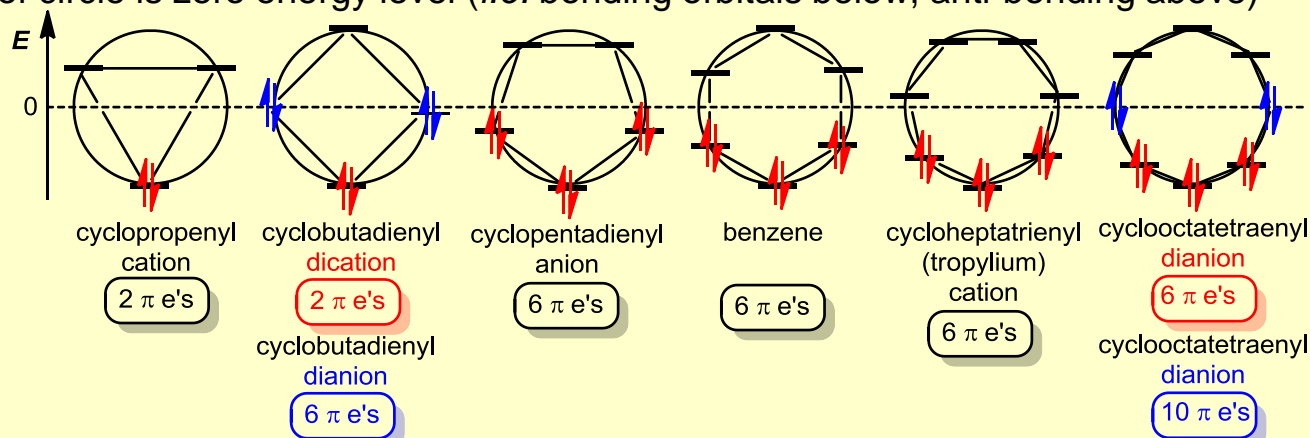
■ Hückel *Z. Phys.* **1931**, *70*, 204; **empirical rule for aromaticity**

- For compounds which are planar & have a contiguous, cyclic array of p-orbitals perpendicular to plane of ring:
  - Those with  $4n+2$  p electrons display special stabilisation: *i.e.* **aromatic**
  - Those with  $4n$  p electrons display special instability: *i.e.* **anti-aromatic**



■ Frost & Musulin *J. Chem. Phys.* **1953**, *21*, 572 ([DOI](#)); **graphical device for constructing MO energy diagrams**

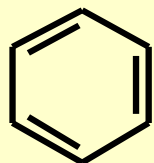
- Draw appropriate regular polygon within a circle (with atoms touching circumference)
- Ensure one atom is at lowest point  $\rightarrow$  ring atom positions represent energy levels
- Centre of circle is zero energy level (*i.e.* bonding orbitals below, anti-bonding above)



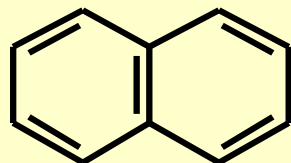
## Other Aromatic Compounds - Resonance Energies

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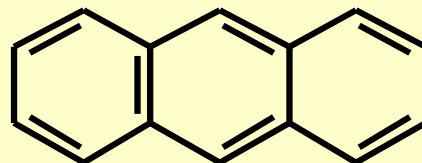
- The anomalous stability of benzene and other aromatic compounds is due primarily to the resonance stabilization gained by the highly symmetric electron delocalisation
  - A large energy input is required to disrupt the cyclically-arranged electrons
  - Consequently, aromatic compounds react slowly with electrophiles
- Other **carboaromatic** compounds show similar resonance energies (in kJ/mol)



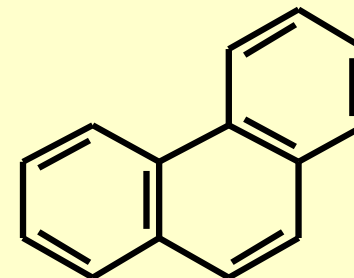
151.2



252

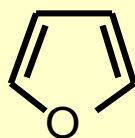


342.8

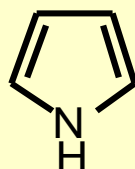


392.1

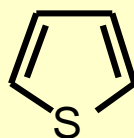
- **Heteroaromatic** compounds usually have lower resonance energies (in kJ/mol)



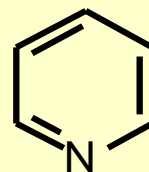
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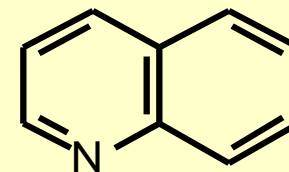
90.1



122



117.2



222

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