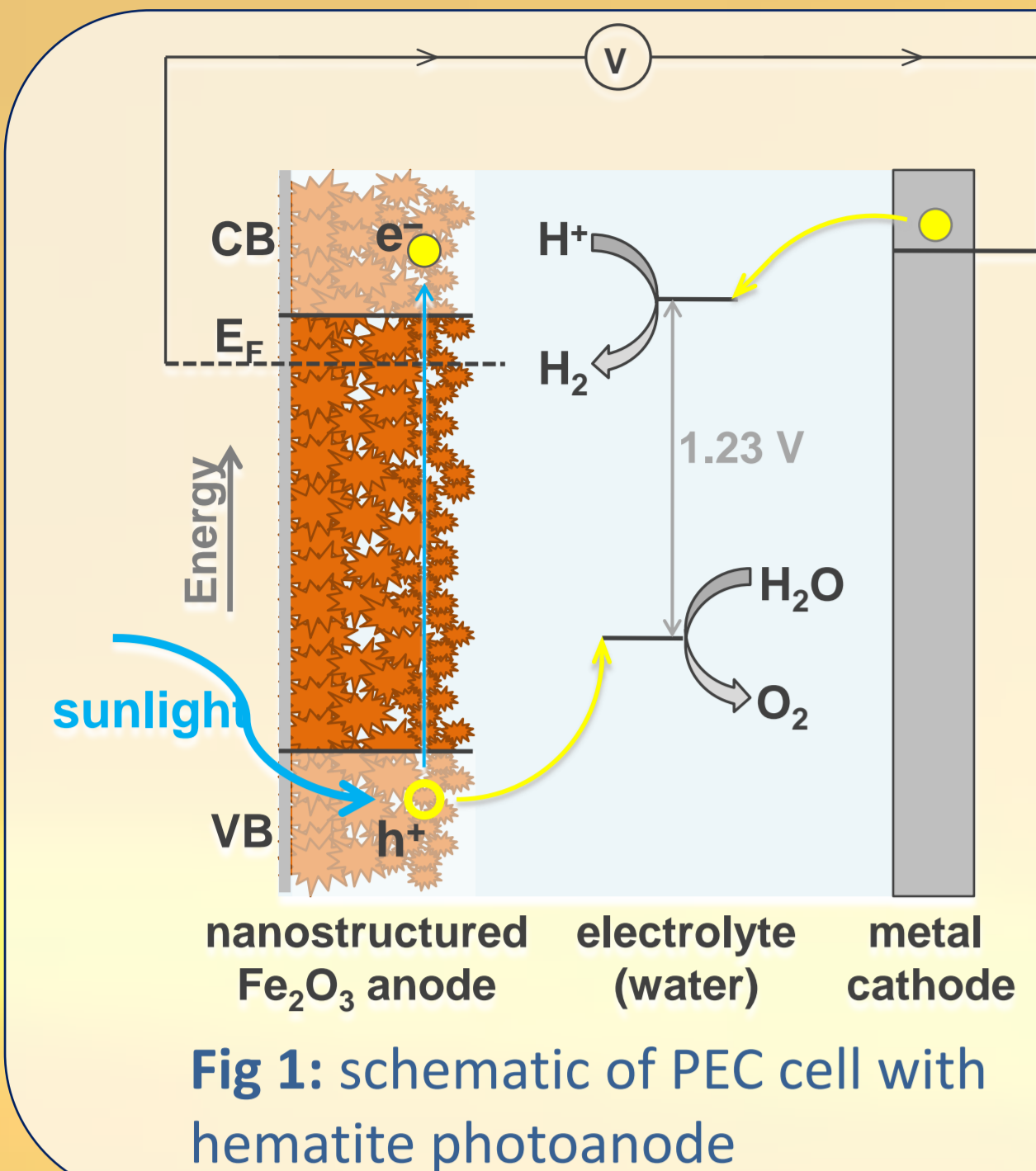


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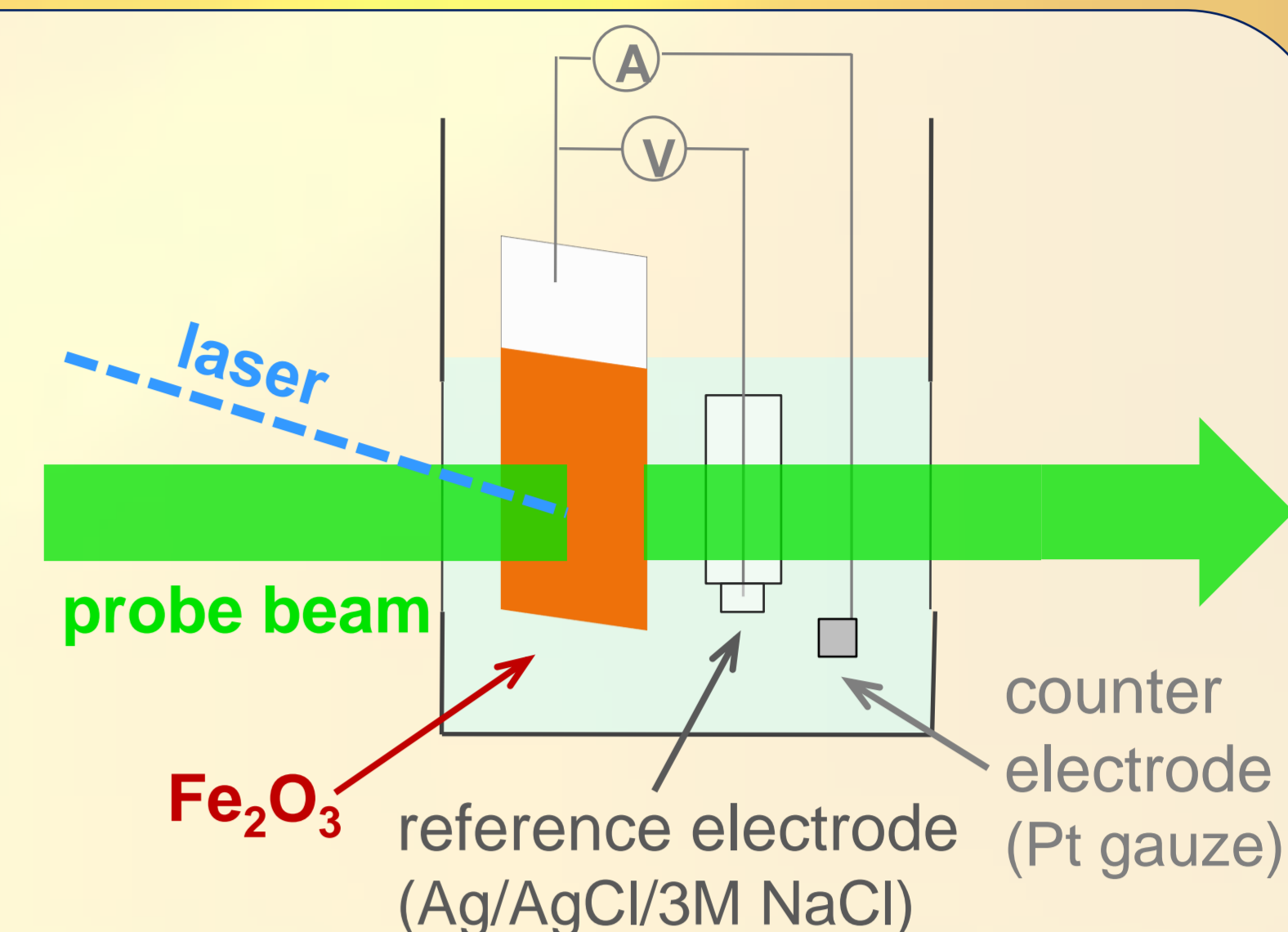


Introduction

Hematite ($\alpha\text{-Fe}_2\text{O}_3$) has several advantages as a photoanode material for solar water-splitting photoelectrochemical (PEC) cells: it absorbs strongly in the visible region, with a bandgap of ~ 2.1 eV (590 nm); the valence band edge is appropriate for water oxidation;¹ it is stable under water photolysis conditions, and is formed from non-toxic, abundant and cheap materials.

However, efficiencies are thought to be limited by poor charge-transport properties,² rapid recombination^{3, 4} and slow charge transfer kinetics at the semiconductor-liquid junction.⁵ Positive applied potential is necessary to reduce the electron-hole recombination rate such that water oxidation can occur,⁶ and because the conduction band edge is positive of the H^+/H_2 redox potential.¹

We use **transient absorption spectroscopy** (TAS, a pump-probe technique) and **transient photocurrent** (TPC) measurements of hematite photoanodes in a complete PEC cell to probe photogenerated holes and electrons respectively.



Correlation of photogenerated hole population with photocurrent

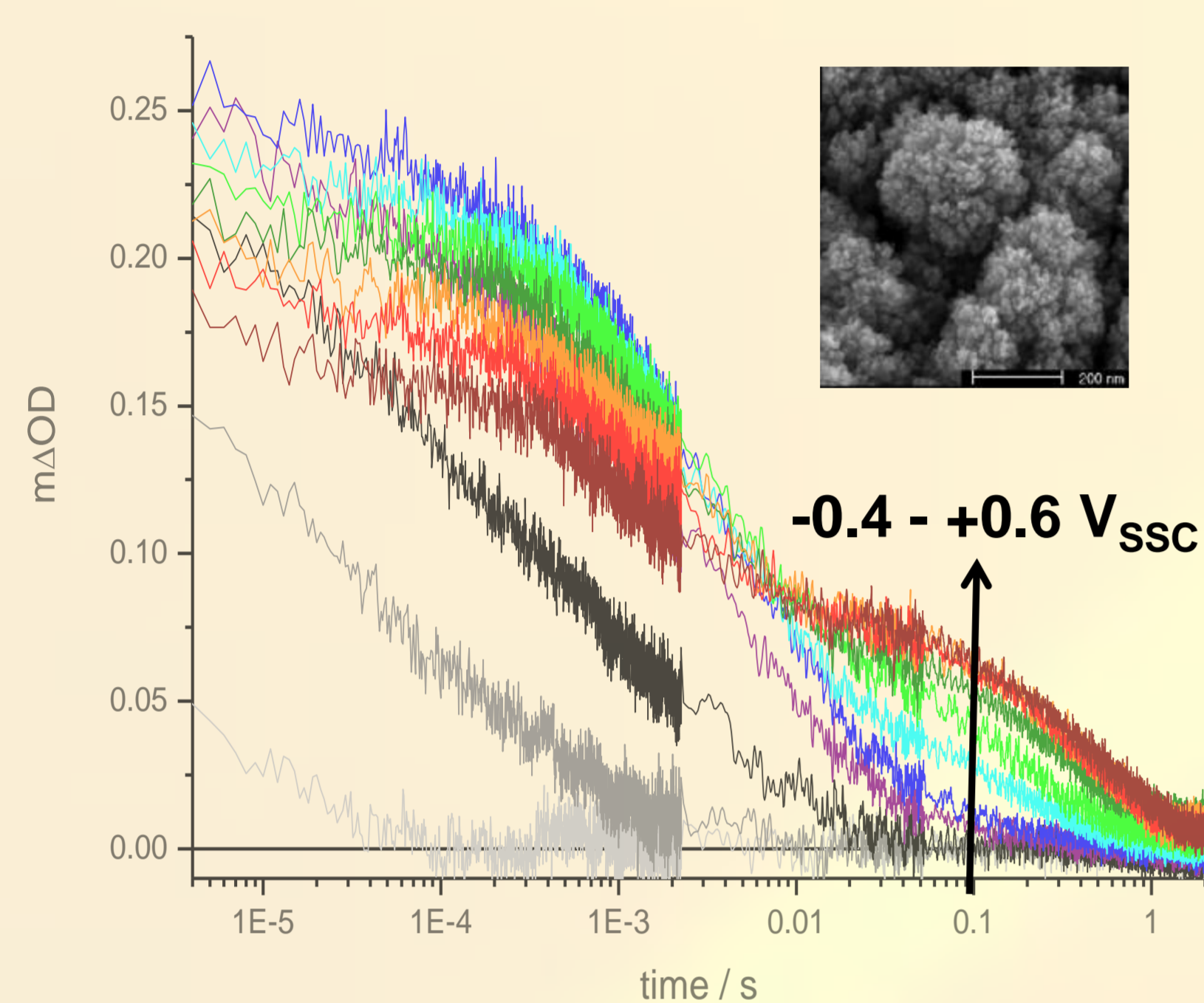


Fig 3: TA decays of the photo-hole (EE 355 nm excitation, probed at 650 nm) as a function of applied bias for Si-doped APCVD hematite photoanodes (SEM⁷ inset)

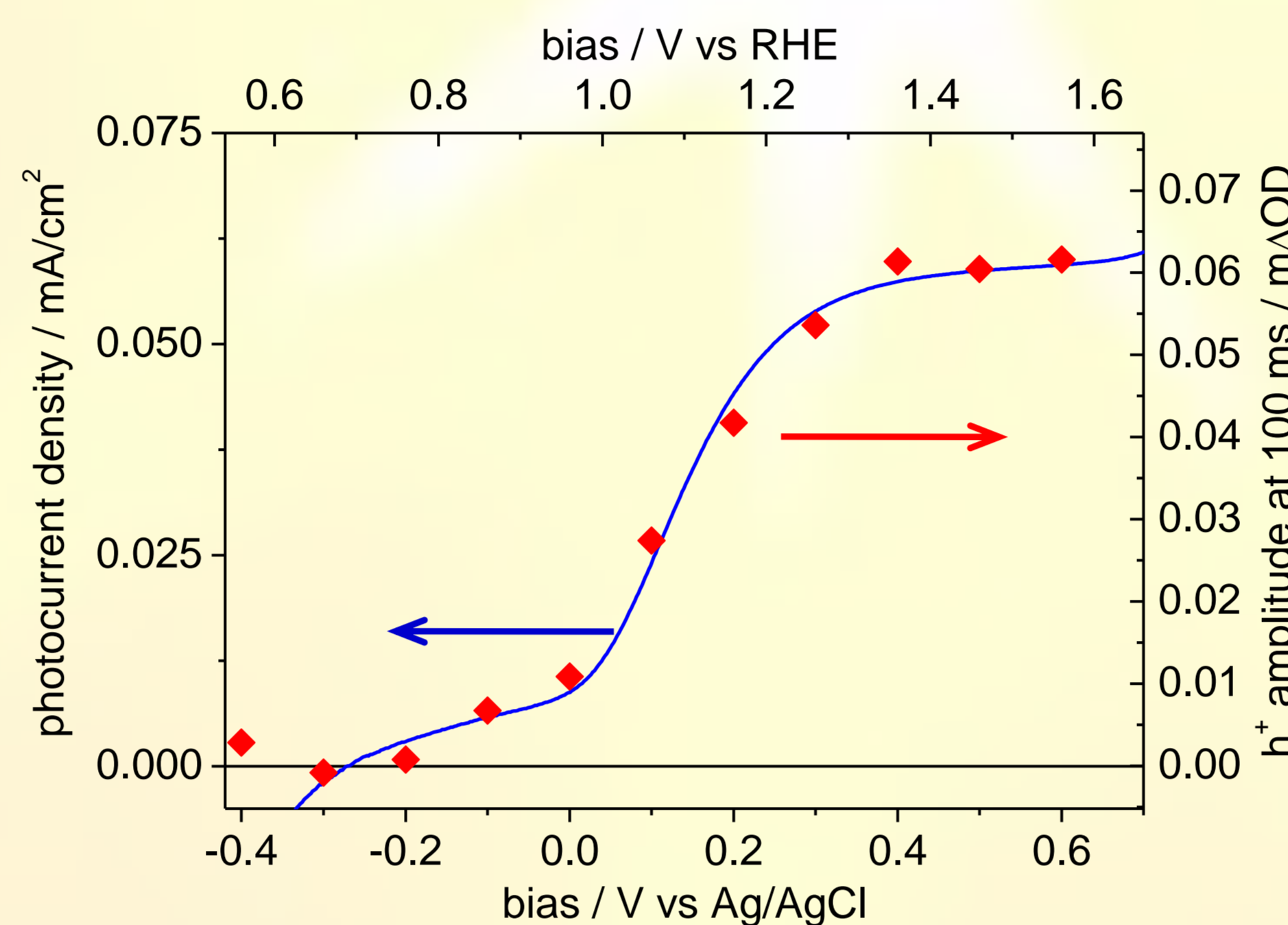


Fig 4: Correlation of long-lived hole signal amplitude (at 100 ms) with photocurrent

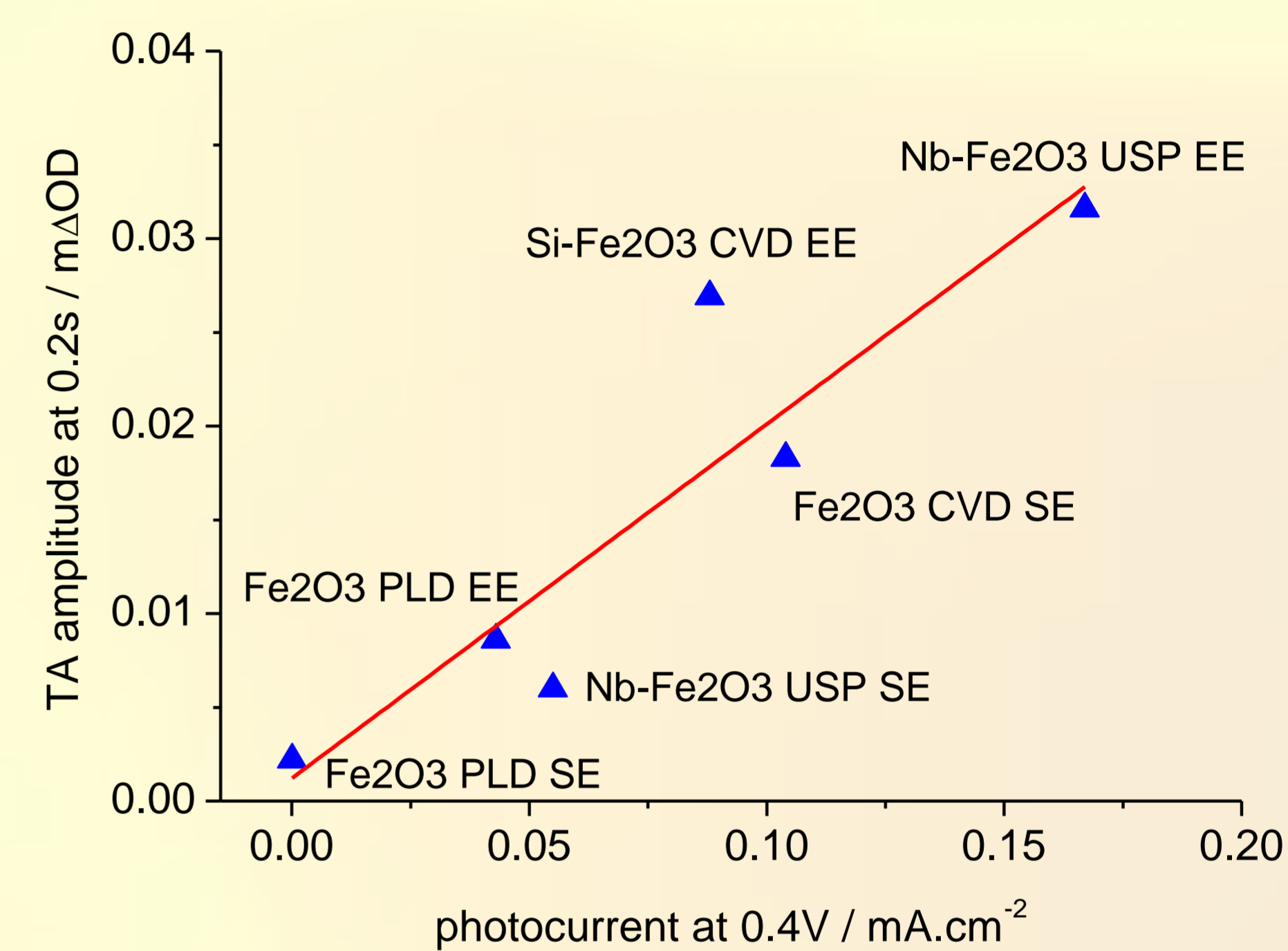


Fig 5: Comparison of long-lived hole population with photocurrent at +0.4 V_{SSC} for various different hematite photoanodes

- The photogenerated holes are monitored by transient absorption spectroscopy (TAS)
- Decay dynamics are strongly dependent on applied electrical bias (Fig 3)
- The fast phase of the TA photo-hole decay (1 μs – 20 ms) is associated with non-geminate electron-hole recombination
- Increasing positive bias reduces the background electron density and increases band-bending, so increasing hole lifetime
- The slow phase (>20 ms) of the TA photo-hole decay is associated with water oxidation

- Water oxidation occurs on a timescale of 100s ms to seconds on hematite \rightarrow very long-lived holes required - positive applied bias necessary
- The timescale of water oxidation is independent of applied bias (Fig 3)
- There is a strong, quantitative correlation between the amplitude of the long-lived photo-hole signal and the photocurrent, as a function of applied bias (Fig 4)
- This correlation is general for different types of hematite (doped, undoped, nanostructured, solid; Fig 5), and also for nanoporous TiO_2 – may be generally true for metal oxide photoanodes

Comparison of transient photocurrent (e^-) and transient absorption (h^+) decays

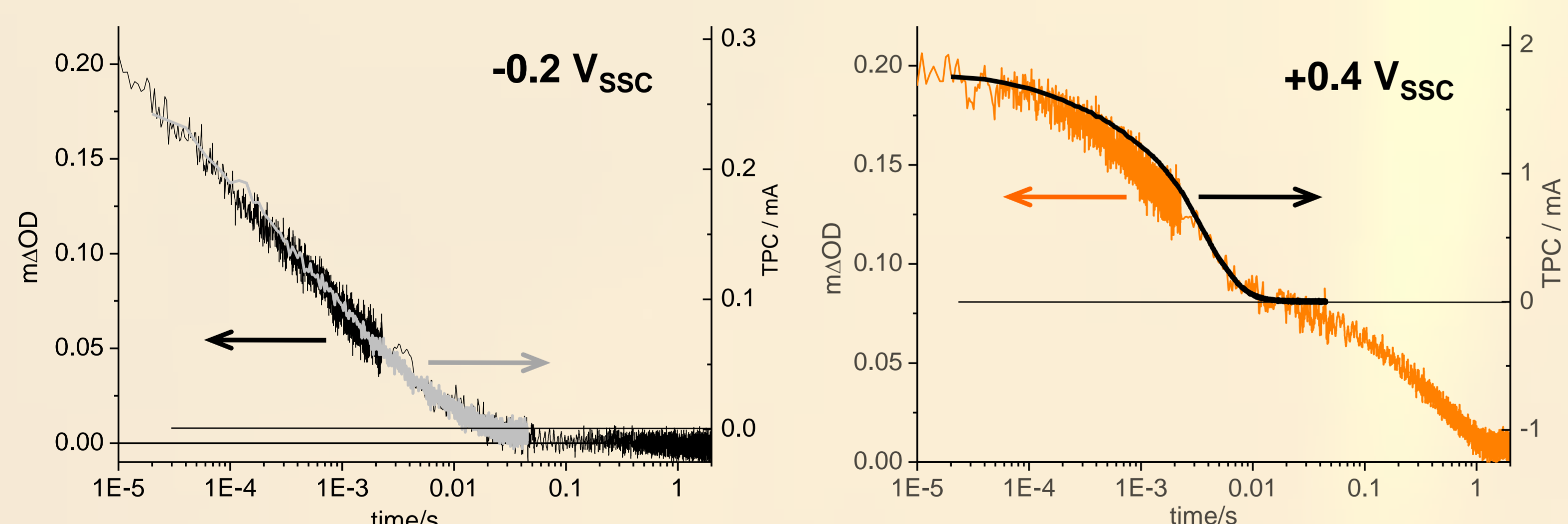


Fig 6: TPC decays (e^-) overlaid on corresponding TA decays (h^+) (355 nm EE excitation, APCVD Si- Fe_2O_3)

- Photogenerated electrons are monitored by transient photocurrent (TPC)
- TPC (e^-) signals very similar to fast phase of transient absorption (h^+) decay (Fig 6)
- Both TPC and fast phase of TA decay dominated by electron-hole recombination \rightarrow transient absorption fast phase and TPC decays have the same shape
- Electron extraction and electron-hole recombination complete by ~ 20 ms \rightarrow electron extraction >2 orders of magnitude faster than water oxidation
- Long-lived hole population (TA slow phase amplitude) limited by electron-hole recombination⁸ at timescales <20 ms
- Significant recombination very likely occurs faster than the timescale of our TA measurements (i.e. <1 μs), particularly at negative bias

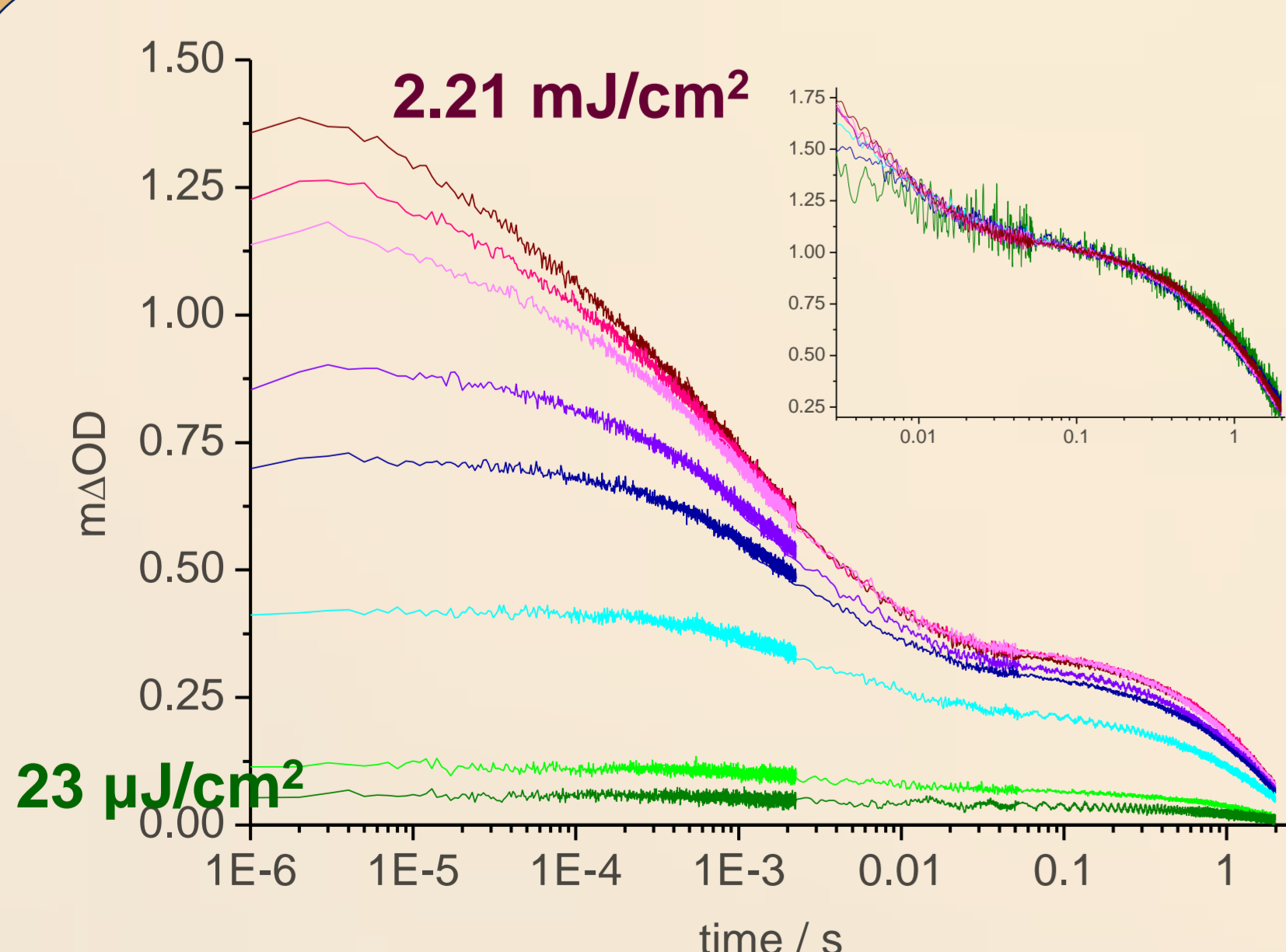


Fig 7: TA h^+ decay as a function of excitation intensity (charge carrier density) at +0.4 V_{SSC} . Inset: normalised slow phase TA decay.

Excitation intensity dependence

- increasing excitation (charge carrier) density increases the TA fast phase decay rate \rightarrow increases recombination rate
- Decay dynamics at two lowest excitation densities almost identical \rightarrow approaching pseudo-first-order recombination
- water oxidation timescale is independent of hole density (excitation intensity; inset Fig 7) \rightarrow RDS of water oxidation mechanism is a single-hole transfer step, to surface-bound or electrolyte water species, i.e. *not* concerted 4-hole oxidation mechanism

Concluding remarks

- Charge carrier dynamics in hematite photoanodes are strongly dependent on electron density (controlled by bias and excitation intensity)
- Electron-hole recombination and electron extraction occur significantly faster than water oxidation; the recombination rate increases with increasing electron density
- There is a strong, quantitative correlation between long-lived hole population and photocurrent – limited by electron-hole recombination
- water oxidation timescale is independent of hole density, indicating RDS is a single-hole transfer, not concerted 4-hole mechanism
- optimisation efforts should concentrate on reducing recombination and/or increasing the rate of electron extraction to the external circuit

Acknowledgements

Piers Barnes and Stephen Dennison for helpful discussions and SocMan Ho Kimura for lab assistance. Michael Grätzel and Kevin Sivula at EPFL and Jinhua Ye for providing materials. Funding from EPSRC is gratefully acknowledged.



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