# **TISSUE FIXATION**

# (paraformaldehyde) January 2010

# **SECTION 1** Summary of health and safety (Do not fill in)

MAJOR DANGERS
These procedures use very toxic chemicals of necessity. They should only be performed by competent persons.

Paraformaldehyde

### **GENERAL DANGERS**

NaOH cause burns

**PPE** 

**CLOTHING** Lab coat **GLOVES** Nitrile **EYE PROTECTION** 

**MASK** FIRST AID

Splash...Wash immediately remove contaminated clothing, wash before reuse Ingestion...Seek medical help

### SPILLAGE PROCEDURES

Minor...Wash down with water

Major...Adsorb onto spill kit. Remove via contractor

### WASTE DISPOSAL

Dilute / small quantity dilute to drain

Concentrated / large quantity via contractor

# SECTION 2 LIST ALL RELEVENT CHEMICALS (ALL VERY TOXIC, TOXIC, HARMFUL OR IRRITANT)

Cat No.	CHEMICAL	MAX QTY (1)		STATE (2)		PROCESS (3)		R PHRASE		TOTAL
S 8045 (s)	<b>Stock Solution</b>									
	Sodium Hydroxide	В	2	A	1	С	3	35	M	6M=GLP
	Working Solutions									
S 8045 (s)	1M Sodium Hydroxide	A	1	A	1	A	1	35	M	1M=GLP
P 6148 (s)	Paraformaldehyde	В	2	C	3	С	3	11-45-46-23-21/22-42/43	X	18X
22290 (m)	Saturated Picric Acid	A	1	A	1	В	2	22	M	2M=GLP

#### Notes

1) MAX QTY A..<1gm (ml) B...1-100gm (ml) C...>100gm (ml)

2) STATE A..dense solid, non volatile liquid

B..dusty solid, volatile liquid

C..powder, highly volatile liquid, aerosol, fume, gas or substance adsorbed through skin

3) **PROCESS** A.. Predominantly closed ie operator not exposed (eg stoppered vessel)

B..partially closed (eg a conical flask)

C..open ie operator exposed to chemical (eg weighing out into a weighing boat)

4) Do **NOT** fill in the thin columns or the TOTAL column.

NB. The list of catalogue numbers represents the usual items used, but may be replaced in practice by equivalent alternatives from a variety of suppliers ((s) - Sigma, (r) - Roche, (bd) - Becton Dickinson, (i) - Invitrogen, (m) - VWR International, (p) - Promega)

#### **SECTION3** DESCRIBE PROCEDURE

### **Fixation of Tissue**

Formaldehyde fixative (freshly made from Paraformaldehyde – required particularly for immunofluorescence) is used in varying concentrations upto 4% to fix biological material for a variety of techniques (light & electron microscopy)

#### 1. Chemicals:

Paraformaldehyde

0.1M phosphate buffer saline pH 7.2-7.6(Oxoid - Dulbecco "A" PBS tablets Cat. No. BR14a)

1M NaOH solution (Stock = 40 gm/100 ml)

Saturated Solution Picric Acid (option for Zambonis Fixative)

### 2. Fixative preparation: (per 100ml) example 2% paraformaldeyde Solution in PBS

- 1) Heat 30ml of distilled water (in conical flask) in microwave oven to near boiling point and then transfer onto magnetic stirrer in the fume cupboard.
- 2) 2g paraformaldehyde are weighed in a weighing boat on chemical balance in the laboratory, (this cannot be performed in the fume hood due to the air flow blowing the light powder about) and added to the distilled water in the fume hood.
- 3) After stirring for 5 minutes add 1 drop of 1M NaOH to clear the solution. Add 1 phosphate buffered saline tablet, dissolve and make up to 100ml with distilled water.

**Variation**: Zamboni's fixative. (2% paraformaldehyde in PBS with picric acid) Preparation as above with the addition of 200µl of a saturated solution of Picric acid after step 3

#### 3. Method:

Material to be fixed is place into the fixative solution in a closed container (bottle/flask) for required amount of time.

The fixative is then poured off and disposed down the fume cupboard sink and replaced with PBS buffer.

The material is then ready for further processing as required.

# 4. Waste

Small quantities of fixative solution may be disposed down the fume hood sink, greatly diluted with cold running water.